## Q.P. Code: 75753

Total Marks 60 Time 2 ½ hrs

## Instructions

<ol> <li>All questions are compulsory.</li> <li>All questions carry equal marks.</li> <li>Draw diagrams where ever neces</li> </ol>	sary.	
Q. 1. a) Using structures, show the formation	on of a peptide bond. Enumerate the pi	roperties of the
peptide bond.	·	06
b) Describe the structural features of	the alpha helix and the beta sheet. Cor	nments on domain
and motifs.		06
	OR	
Q.1 a) Describe the structure of the collage	en molecule , explaining the special fea	tures of its structure
		08
<ul> <li>a) Write briefly on collagen cross link</li> </ul>	<b>C.</b>	04
Q.2 a) Discuss inter molecular interaction	with reference to potential energy diag	ram and describe
deviations from Hooks law.		06
b) what are protein folding rules? Wha	it is propensity	06
	OR	
	ts. Why it is considered to be an entrop	y effect rather than
entalpic ?	•	06
b) Explain the donar acceptor pheno	menon in the interactions between wat	er and a polypeptide
and role of H bonds		06
Q.3. a) Discuss macroscopic and microscopic liands binding equilibria		06
b) Explain Scatechard plot and its applications		06
	OR	
Q.3. a)What are helix coil transition? What is Hill constant		06
b)Comments on the Random coils		06
Q.4 a) Chargaff's rules: A student experime	•	•
thymine content. Calculate the percentage of A G and C for the sample.  b.). Discuss: i) Importance of G:C content on DNA stability ii) DNA melting of		03
	of base sequence (Hint: steepness of n	
die content in morting as a function	of base sequence (fine, steephess of fi	09
	OR	
Q.4. Discuss the forms of DNA in relation	·	
helices. Compare the three classic doub	ole nelices A, B and Z.	12
Q.5 Write notes on <i>any three</i>		12
a) Disulphide bridges and protein stability	d) structure of hemoglobin	
b) Interionic attraction and ion-atmosphere	e e) Ramchandran plot	
c) Co-operative interaction	f) Urea as denaturing agent	