**QP Code: 76078** 

(3 Hours)

[ Total Marks: 100

N.B.: (1) All questions are compulsory

- (2) Figures to the right indicate full marks.
- 1. Attempt any five of the following:

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- (a) Explain:
  - (i) t-butoxide ion is a strong base but a weaker nucleophile than hydroxide ion.
  - (ii) Vinyl halides do not undergo  $S_{N^1}$  and  $S_{N^2}$  reactions.
- (b) The pKa value of o-hydroxy benzoic acid is lower than the meta and para isomer. Explain.
- (c) Predict the product and give the mechanism:

$$CH_3COOC_2H_5 + C_6H_5MgBr \xrightarrow{dryTHF/H_3O^+} ?$$
(excess)

(d) Complete the reaction and give the mechanism:

$$CH_3CH_2CH_2COOC_2H_5 \xrightarrow{\text{(i) Na,xylene,}\Delta}$$
?

(e) Complete the reaction and give its mechanism:

- (f) What is Friedel Craft acylation? Give the mechanism of the reaction
- (g) Write a note on Retro Diels Alder reaction.
- (h) How will you distinguish between cis and trans alkene on the basis of NMR spectroscopy?
- 2. (a) Answer the following:-

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- (i) Explain the mechanism and stereochemistry of  $S_{N^{i}}$  reactions.
- (ii) What are ambident nucleophiles? Give two examples.

OR

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(a) Show the R<sub>i</sub> and S<sub>i</sub> faces of the following compound. Give the product obtained when the S<sub>i</sub> face of the compound is treated with HCN.

- (b) How is the mechanistic pathway of ester hydrolysis determined by isotopic labelling? Explain in detail, the A<sub>AC</sub><sup>2</sup> mechanism of ester hydrolysis.
- c) Give an account of stereochemistry of ansa compounds. Give a methodology for assigning stereochemical descriptors to these compounds. Explain with examples.

## OR

(c) Assign configurational descriptors to the following compounds :

(A)  $\frac{N_{02}}{C_{00}H^{N_{02}}}$  (B)  $\frac{N_{02}}{C_{H3}}$  COOH CH3

To which class of compounds does (A) and (B) belong? Discuss the stereochemistry of any one of the above.

3. (a) Complete the following reactions and give their mechanism:

(i) CH3 + BrCH2COOC2H5 Zn >?

## OR

(a) Explain kinetic and thermodynamic control in alkylation reaction of an unsymmetrical ketone using a suitable base.

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(b) Complete and name the following reactions:

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(iii) 
$$CH_3 CH_2 COO C_2H_5 + \frac{0.00C_2H_5}{0.00C_2H_5} \xrightarrow{0.00C_2H_5} \frac{0.00C_2H_5}{0.00C_2H_5} \xrightarrow{0.00C_2H_5} \frac{0.00C_2H_5}{0.00C_2H_5}$$

(c) Give the mechanism and one application of Mc Murry's coupling reaction.

OR

(c) Give the mechanism and one application of Perkin reaction.

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4. (a) Predict the product and name the reaction :-

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$$\frac{Zn/Hg}{HcI} \rightarrow ?$$

$$\begin{array}{c} \text{(ii)} & \text{(CH3)} \\ \text{(CH3)} & \text{(CH-OH)} + \text{(CH3)} \\ \text{(CH3)} & \text{(CH3)} \end{array}$$

OR

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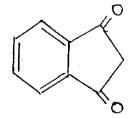
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$$\xrightarrow{Cf_3COOH}$$

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(b) Complete the following reactions:-





$$SeO_2$$

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$$\frac{\text{cooh}}{\text{No}_2} \xrightarrow{\text{HN}_3/\text{H}_2\text{So}_4} ?$$

$$V) CH_{3} \xrightarrow{NH_{2}} CH - COOH \xrightarrow{H_{2}O/\Delta} ?$$

(c) Give any two applications of NaBH<sub>4</sub> reagent and give the mechanism of any one of them.

OR

- (c) Answer the following:
  - (i) Give the synthetic applications of MnO<sub>2</sub> reagent.
  - (ii) Discuss the mechanism of Fries rearrangement.
- 5. (a) What is meant by the term 'Chemical shift'? Discuss the factors affecting chemical shift.

OR

- (a) (i) Explain McLafferty rearrangement in mass spectrometry.
  - (ii) Give the fragmentation pattern of benzoic acid.
- (b) An organic compound having molecular formula C<sub>4</sub>H<sub>5</sub>NO<sub>2</sub> shows following spectral data:-

IR  $(cm^{-1})$ : 2250(m), 1740 (s)

PMR (δ ppm) : 3.8 (3H, singlet), 3.5 (2H, singlet)

Mass sectrum (m/z): 99, 73, 59

Deduce the structure of the compound.

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(c) (i) How will you distinguish between the following pairs on the basis of IR

- spectroscopy?
  - HC-CH2-CH
- and I

Y,



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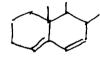
(ii) Explain finger print region in IR spectroscopy.

OR

(c) (i) Calculate  $\chi_{\,\text{max}}$  for the following :

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(ii) What is bathochromic shift and hypsochromic shift?