Total marks :60 (2 1/2		Hours)	
NB : All	I questions are compulsory		
	l questions carry equal marks	7000	
	Attempt any two of the following	[12]	
i	The average molecular weight of an amino acid residue is 120. The average density of a soluble protein is 1.33 g/cm ³ . Calculate (I) the specific volume of an average soluble protein. (II) the weight of a single molecule of a protein containing 270 amino acids, and (III) the volume occupied by a single molecule of this protein.	4 4 4 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6	
ii	Discuss about frequently used methods of cell disruption for extraction of biomolecules.		
iii	How would you precipitate proteins using organic solvents?	72/20	
iv	Write short note on any three methods of qualitative analysis of carbohydrates.		
Q.2	Attempt any two of the following	12	
i	Derive equation of Michaelis Menten reaction Comment on Haldane relationship in kinetics of enzyme catalysed reactions		
ii	Explain thermodynamic aspects of the enzyme-catalysed reaction with reference to transition state.		
iii	Explain the mechanism of enzyme catalysis of Ribonuclease A.		
iv	Describe briefly enzyme regulation mediated through reversible covalent modifications. Elaborate your answer with specific example.		
0.3	Attempt any two of the following	12	
i	Describe the signaling systems which control luminescence in <i>Vibrio harveyi</i> .		
ii	The availability of ammonia regulates the Ntr regulon in enteric bacteria, Justify.		
iii	Comment on the catabolite repression system of <i>Escherichia coli</i> that does not involve camp		
iv	Explain signal transduction during chemotaxis in Salmonella typhimurium.		
Q.4 i	Attempt any two of the following With the help of structure and enzymes describe degradation of naphthalene to catechol.	12	
ii	Give an account of oxidation of propane by microorganisms.		
iii	Describe degradation of aromatic compounds through anaerobic photometabolism and nitrate respiration.		
o iv	Discuss degradation of aliphatic hydrocarbon through cytochrome P450 dependent		
22 25 25 25 25 25 25 25 25 25 25 25 25 2	enzymatic system.		
Q.5(a)	Explain (any two)	04	
9 6 6 15 IS	Basis of separation of biomolecules in gel filtration technique		
o Sii	Transition state analogues		
o dii	Competence in Bacillus subtilis		
iv	Diterminal oxidation of hydrocarbons		
5 6 65 6	& & X X & & X X X X X X X X X X X X X X		

TURN OVER

Q.5(b)	Give significance/function of (any two)	0.
i	Edman degradation	
ii	Mixed inhibitor	300
iii	RapA phosphatase	3
iv	Rubridoxin	
		70
Q.5(c)	Give one example of (any four)	04
i	Stationary phases used in HPLC	
ii	Chromatography derivatization agents	2/2
iii	Irreversible enzyme inhibitors	, C
iv	Major types of drug design	3
V	Genes induced by bvg gene product	XX C
vi	Cytoplasmic histidine kinase	9
vii	Products of phenanthrene degradation	
viii	End product of orthocleavage of catechol by microorganisms	
